Effect of Roof Material on Water Quality for Rainwater Harvesting Systems – Progress Report to the Texas Water Development Board (TWDB)

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> **Prepared for:** Texas Water Development Board

The main objective of this research is to provide recommendations to the rainwater harvesting community in Texas regarding the selection of roofing material for rainwater harvesting and to support these recommendations with scientific data. In Task 1, we have identified roofing materials that are commonly used in Texas and those that are commonly recommended in Texas for rainwater harvesting. In Task 2, we are examining the quality of harvested rainwater from pilot-scale roofs constructed with traditional materials (i.e., asphalt fiberglass shingles, galvanized metal, concrete tiles) and alternative materials (i.e., green and cool roofs). In Task 3, we are examining the quality of harvested rainwater from two existing full-scale residential roofs. In Task 4, we will prepare and submit a final report for the research. The following provides a summary of our progress to date for Tasks 1 through 3.

Preliminary Data

Before initiating this project with the TWDB, preliminary data on metal concentrations in harvested rainwater were gathered from an existing metal roof and an existing asphalt fiberglass shingle roof. Dechlorinated tapwater was sprinkled on the roofs at the field sites, and the concentrations of arsenic, zinc, cadmium, and lead were measured (Table 1). As expected, the metal roof showed the highest concentration of metals among the sites tested.

Roof Type	Metal concentrations (µg/L)				
	Arsenic	Zinc	Cadmium	Lead	
Blank (dechlorinated tapwater)	1.1	4.5	ND	1.4	
Asphalt fiberglass shingle	1.2	8.5	ND	10.4	
Metal (Galvalume®)	1.0	25.4	ND	17.1	
ND: not datastad					

Table 1. Metals Concentrations in Runoff from Different Types of Roofing Materials

ND: not detected

Task 1. Survey of Roofing Materials Commonly Used in Texas (complete)

A survey was conducted to determine which residential roofing materials are most commonly used in Texas and what products are used to adhere, seal, or coat roofing materials. To complete this task, contact information for 71 roofing contractors was collected from the National Roofing Contractors Association (NRCA) and the Midwest Roofing Contractors Association (MRCA); this information is summarized in the Appendix (Table A.1). Forty-five percent of the contractors agreed to participate, yielding a total of 23 residential and 9 commercial roofing contractors who participated in the survey. A brief summary of the survey questions and answers are provided in the Appendix (Table A.2). All commercial and residential roofing contractors confirmed that self-adhesive asphalt fiberglass shingles are the most commonly used residential roofing materials in Texas, being used on more than 80% of residential roofs (Jason Wright, personal communication, 2008). The second most commonly used residential roofing material used in Texas is a type of metal roof called Galvalume®, which is usually fastened with nails. In addition to shingles and metal roofs, it was also reported that concrete roofing tiles are used in

Texas. When asked what roofing materials should be recommended for rainwater harvesting, more than 80% of the contractors said that metal roofs should be used.

To conduct a more thorough investigation of the chemical composition of each roofing material, several material safety data sheets (MSDS) were retrieved from manufacturers that were recommended by commercial and residential contractors. According to the MSDS by Tamko, asphalt fiberglass shingles contain (by weight) <30% asphalt, <65% limestone, <40% mineral granules, <8% fiberglass, and <2.4% formaldehyde (Tamko 2007). This chemical composition is comparable to that listed in the GAF-Elk MSDS, which states that fiberglass shingles contain (by weight) 10-30% asphalt, 25-45% limestone, 20-45% granules, and a fiberglass mat (1-3%) (GAF-Elk 2008). In the toxicological information section of the Tamko MSDS, it is reported that shingles may contain small amounts of Polycyclic Aromatic Hydrocarbons (PAHs): some of these compounds have been classified as carcinogenic (Barone et al. 1996). In particular, it is mentioned that Benzo(a)pyrene has been identified in asphalt fumes. In Tasks 2 and 3, we will be analyzing harvested rainwater samples to determine if they contain selected volatile and semi-volatile organic compounds.

According to the MSDS by Dofasco, Galvalume® sheets contain (by weight) approximately 95% iron, <1.65% manganese, <1.1% chromium, and <0.12% nickel (Dofasco 2007). A variation of these chemical compositions is reported by other manufacturers; BlueScope Steel reports that Galvalume® sheets contain (by weight) 1-10% zinc, 1-10% aluminum, and the remainder is composed of iron (BlueScope Steel 2003); the United States Steel Corporation (USS) reports that Galvalume® sheets contain (by weight) <92% iron and a variety of alloying elements, including aluminum, copper, silicon, sulfur, and manganese, with <1.15% each (USS 2004). These three manufacturers mention that chromium is used as a metallic coating for surface treatment. As a result, it is possible that Galvalume® roofs may leach several types of metals. Thus, in Tasks 2 and 3, we will be analyzing harvested rainwater samples to determine selected metal concentrations.

A combination of organic and metallic constituents is reported in the chemical composition of concrete tiles. According to the MSDS by MonierLifetile Manufacturing, concrete tile is composed of (by weight) 20-30% cement, 50-60% sand and aggregate, 0-5% limestone, and 0-8% acrylic polymer; in addition, concrete tiles contain a mixture of metal pigments, including cobalt, chromium, and titanium, each ranging between 0-3%. (MonierLifetile 1999). Thus, in Task 2, we will be analyzing harvested rainwater samples to determine selected organic and metal concentrations from concrete tile roofs.

Task 2. Test Roofs (currently underway and on schedule)

Based on the results of Task 1, three roofing materials were selected for the test roofs to be installed at the Lady Bird Johnson Wildflower Center: GAF-Elk's asphalt fiberglass shingle, Berridge's Galvalume® standing seam metal, and MonierLifetile's concrete tile. Three wooden frames were constructed to hold a 4 ft by 8 ft plywood sheet that will support the roofing material. To simulate residential roofs, Austin contractors recommended that the pilot roofs have a pitch of 4:12 (Wilson Ralph, personal communication, 2009 and Harold Peterson, personal communication, 2009). As a result, the plywood boards were angled at 18.4 degrees. The frames are approximately 5-ft high with a 2-ft end for rainwater catchment (Figure 1).



Figure 1. Frames for Pilot-Scale Roofs

The three wooden frames were installed at the Lady Bird Johnson Wildflower Center, and they will be secured with concrete footings. D.L. Philips Construction Co., Inc. (Austin, Texas) has volunteered to install the three roofing materials on the frames, at no charge, by the end of March 2009 (David Philips, personal communication, 2009). Once the construction of the three roofs has been completed, we will construct a guttering system similar to that utilized in Task 3 (see **Error! Reference source not found.**). In addition, one of the six green roofs that are already installed at the Lady Bird Johnson Wildflower Center will be included in these pilot-scale experiments. Jeffrey Stump, an undergraduate student at the University of Texas at Austin (and the recipient of an Undergraduate Research Fellowship from the University), will also collect rainwater from a cool roof that is already installed onsite.

Task 3. Field Sampling (currently underway and on schedule)

The purpose of this task is to examine the quality of rainwater harvested from existing residential roofs in the Austin, Texas area. Two full-scale roofs were sampled: a twelve-year-old metal roof (Galvalume®) on a single-story residence and a five-year-old asphalt fiberglass shingle roof on a two-story residence. These sites allow us to investigate the quality of rainwater harvested from aged, full-scale roofs.

To collect the rainwater, the base of each roof was equipped with a simple sampling device that was inserted into the gutter (Figure 2). This insert consists of a clean 3-inch diameter PVC pipe (potable quality) cut lengthwise in half and fitted with end caps. Three-quarter-inch PVC pipe was used to direct rainwater collected from the sampling insert to a passive collection system (**Error! Reference source not found.**) that consists of a two-liter (L) tank to collect the "first flush" and two 10-L tanks in series to collect volume samples after the first flush.



Figure 2. Sampler Gutter Insert



Figure 3. Sampler Design

It is recommended that the first flush divert a *minimum* of ten gallons for every 1,000 square feet of true collection surface area (TWDB 2005). Since each roof collection area used in this task was approximately 50 square feet, we diverted slightly more than half a gallon (two liters) to ensure that we met the minimum recommendation for first flush volume. The tank volumes were determined based on the estimation that one inch of rain will result in one half-gallon of collected water for every square foot of roof footprint area (TWDB 2005). Based on our collection data, the average rainfall in the Austin area has been approximately one half-inch for

the majority of events in the past year. Therefore, we estimated that the system could collect approximately 13.5 gallons (about 50 liters) for a full rain event.

In addition, each site was equipped with a separate sampler to collect ambient rainwater that does not come into contact with the roof. This sampler (Figure 4) consisted of an 18-inch diameter polyethylene funnel attached to a 10-L polypropylene collection container.



Figure 4. Ambient Sampler

Each of the two residential roofs was sampled for three rainfall events (2/9/2009, 2/11/2009, and 3/11/2009). Between events, each sampling tank was thoroughly washed with soap, rinsed with deionized water, and autoclaved. The remaining pieces of the field sampler (e.g., PVC piping and funnel) were scrubbed and rinsed with deionized water on site.

Table 2 summarizes the analytical methods that are being used, and Table 3 lists the preservation methods and storage times for each type of sample.

Most of the analyses have been completed for the first two rain events; the remaining measurements for the first two rain events and the measurements for the third rain event (3/11/09) are currently being performed. The results for the analyses conducted thus far for the first two rain events are presented as follows: pH (Error! Reference source not found.), conductivity (Error! Reference source not found.), turbidity (Figure 7), total solids (Figure 8), total coliform (Error! Reference source not found.), fecal coliform (Error! Reference source not found.), nitrate (Error! Reference source not found.), and selected synthetic organic contaminants (Table 4). Samples for metals have been preserved with nitric acid and are being stored at 4°C until

analysis. In addition, DNA has been extracted from all samples and is being stored at -80°C for future microbial community analysis (if additional funding becomes available).

Table 2. Analytical Methods						
Parameter	Method Title	Meter/Method Type	Source			
pН	pH Value	Potentiometry	Standard			
			Methods (1998)			
Conductivity	Conductivity	Copenhagen Conductivity Meter	Copenhagen			
	$(\mu S/cm)$		Radiometer			
Turbidity	Turbidity (FTU)	Hach Turbidity Meter Model 2300	Hach (2003)			
Total Solids		Filter and dry	Standard			
			Methods (1998)			
Total	M-Endo Broth		Standard			
Coliform			Methods (1998)			
Fecal	FC Agar		Standard			
Coliform			Methods (1998)			
Nitrate	Chromotropic	Colorimetric	Hach (2003)			
	Acid Method					
Synthetic	Method	(See list in Appendix: Table A.3)	DHL Analytical			
Organic	8260/8270		Laboratories			
Compounds						
DNA	MoBio UltraClear	n Microbial/Soil DNA Isolation Kit	MoBio			
Extraction						

Table 2. Analytical Methods

Table 3. Sample Preservation and Stora	ge
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Measurement	Preservation	Maximum Holding Time
"II	None as suized	Ŭ
рН	None required	N/A
Conductivity	None required	N/A
Turbidity	None required	N/A
Total Solids	None required	N/A
Total Coliform	Store at 4°C	6-8 hours
Fecal Coliform	Store at 4°C	6-8 hours
Nitrate	Acidify to $pH < 2$; store at $4^{\circ}C$	28 days
Synthetic Organic	Acidify to $pH < 2$; store at	7 days
Compounds	4°C	
DNA extraction	Refrigerate	2 days
	(-20°C after extraction)	(indefinitely)

N/A: not applicable; analysis is conducted immediately.



Figure 5. pH of Harvested Rainwater from a Metal Roof and a Shingle Roof



Figure 6. Conductivity of Harvested Rainwater from a Metal Roof and a Shingle Roof



Figure 7. Turbidity of Harvested Rainwater from a Metal Roof and a Shingle



Figure 8. Total Solids of Harvested Rainwater from a Metal Roof and a Shingle Roof



Figure 9. Total Coliform of Harvested Rainwater from a Metal Roof and a Shingle



Figure 10. Fecal Coliform of Harvested Rainwater from a Metal Roof and a Shingle Roof



Figure 11. Nitrate of Harvested Rainwater from a Metal Roof and a Shingle Roof

Contaminant	Roof Type	Date	Concentration (µg/L)
2,4 – Dinitrophenol	Metal	2/9/09	3.12
2,4 – Dinitrophenol	Shingle	2/9/09	2.88
Benzyl alcohol	Shingle	2/9/09	0.200

 Table 4. Synthetic Organic Compounds Detected in First Flush Samples

Error! Reference source not found. presents the pH of the ambient and harvested rainwater from two rain events on the metal and shingle roofs. The pH of rainwater is typically around 5.7 (TWDB 2005), and our data for ambient rain support this. In the case of the metal roof, pH decreased from the ambient rainfall to the harvested rainwater. In the case of the shingle roof, pH increased from the ambient rainfall to the harvested rainwater. The pH ranges from our samples are comparable to the pH of other rainwater studies such as Yaziz et al. (1989) and Simmons et al. (2001).

Error! Reference source not found. presents the conductivity of the ambient and harvested rainwater from two events on the metal and shingle roofs. The conductivity decreased from the first flush through the first and second tanks, with final conductivities that were similar to those of ambient rain. Note that the conductivity of the first flush on the 2/11/09 rain event was lower

than that of the first flush on the 2/9/09 rain event. The conductivity values are within the same range found in the harvested rainwater studies of Yaziz et al. (1989) and Simmons et al. (2001).

Figures 7 and 8 present the turbidity and total solids, respectively, of the ambient and harvested rainwater from two events on the metal and shingle roofs. The turbidity and total solids decreased from the first flush through the first and second tanks, with final values of turbidity and total solids that were close to those of ambient rain. The total solids values in the first flush of the metal roof were higher than the total solids values in the first flush of the shingle roof; this might be attributable to particle retention in the porous surface of the shingle roof as compared to the non-porous surface of the metal roof. The turbidity and total solids values are within the same ranges found in the harvested rainwater studies of Yaziz et al. (1989) and Simmons et al. (2001).

Figures 9 and 10 present the total coliform and fecal coliform counts, respectively, of the ambient and harvested rainwater from two events on the metal and shingle roofs. Each roof site had similar conditions with respect to tree cover to minimize differences in potential fecal contamination (e.g., by birds). The total and fecal coliform counts decreased from the first flush through the first and second tanks. However, even the second tank for both roofing materials always had detectable total and fecal coliform, indicating that treatment is needed prior to potable use. The total and fecal coliform counts were always higher in the rainwater harvested from the shingle roof than from the metal roof; to determine if this is truly due to roofing material differences, rather than site variation, a pilot-scale metal roof and shingle roof will be compared side-by-side at the Lady Bird Johnson Wildflower Center in Task 2.

Figure 11 presents the nitrate concentrations of the ambient and harvested rainwater from two events on the metal roof and from one event on the shingle roof. From the shingle roof and from the 2/11/09 event on the metal roof, the nitrate concentrations decreased from the first flush through the first and second tanks. In the second tanks, the nitrate concentrations were less than those in ambient rain. The data from the 2/9/09 event on the metal roof are being reanalyzed since they show a different trend than do the other data sets. By comparison, a Dutch study looked at the nitrate concentration of rainwater (collected with a glass funnel and bottle) and found concentrations up to 1.02 mg/L NO_3 -N (Van Maanen et al. 2000). Deng (1997), however, found nitrate levels in rainwater up to 420 mg/L in some areas of Florida due to anthropogenic pollution. Thus, the <6 mg/L NO₃-N concentrations found in our harvested rainwater samples are not unexpected.

We tested for a suite of 200 synthetic organic compounds (Appendix, Table A.3) on the first flush samples from the 2/9/09 rain event for the metal and shingle roofs. Only two compounds were detected: benzyl alcohol and 2,4-dinitrophenol, but the concentrations were very low (Table 4).

Samples have been preserved with nitric acid and are being stored at 4 °C for metals analysis. We will likely be measuring copper, lead, nickel, selenium, chromium, and iron.

In addition, we have extracted DNA from each sample. If additional funding becomes available, we will use the extracted DNA to look at the microbial communities that are present in harvested

rainwater. This analysis will be done with terminal restriction fragment length polymorphism (T-RFLP). These data will allow us to compare the composition of the microbial communities between different roofing materials. For example, a metal roof might show reduced microbial community diversity, which would be consistent with the lower numbers of total and fecal coliform observed for the metal roof.

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References

Barone P. M. V. B., Camillo A. Jr., and Galvão D.S. 1996. "Theoretical Approach to Identify Carcinogenic Activity of Polycyclic Aromatic Hydrocarbons." *Physical Review Letters* **72(6)**: 1186-1189.

BlueScope Steel. 2003. Material Safety Data Sheet. Accessed 10 March 2009. http://www.australbrick.com.au/pcms_file/Zincalume_Material_Safety_Data_S_631196312302.pdf.

Deng, Y. 1998. "Determination of Major Inorganic Ions in Rainwater by Capillary Electrophoresis." *Water Research* **38(4)**: 2249-2256.

Dofasco. 2007. Steel Material Safety Data Sheets. Accessed 2 March 2009. http://wcm.pavliks.com/WCMAdmin/Images/wwwbmp-groupcom/Customer_Service_Images/MSDSsheets.pdf.

GAF-Elk. 2008. GAF-Elk Materials Corporation Material Safety Data Sheet MSDS Number: 1002. Accessed 10 March 2009. <u>http://www.gaf.com/Content/Documents/20550.pdf</u>

MonierLifetile. 1999. MSDS-Material Safety Data Sheet. Accessed 2 March 2009. http://www.monierlifetile.com/technicaltools/pdf/MSDS.pdf

Peterson H., 16 March 2009. Personal Communication. Manager-Drury Roofing and Sheet Metal Inc.

Philip D., 10 March 2009. Personal Communication. Manager-D.L. Philips Construction Co. Inc.

Ralph W., 16 March 2009. Personal Communication. Manager-Wilson Roofing Co. Inc.

Simmons G., Hope V., Lewis G., Whitmore J., and Gao W. 2001. "Contamination of Potable Roof-collected Rainwater in Auckland, New Zealand". *Water Research* **35**(6): 1518-1524.

Tamko. 2007. Material Safety Data Sheet (MSDS) Number: T029000. Accessed 2 March 2009. http://www.tamko.com/Portals/0/documents/T029000%20Shingles%20PDF%20Version.pdf

Texas Water Development Board (TWDB). 2005. "The Texas Manual on Rainwater Harvesting".

United States Steel Corporation. 2004. United States Steel Corporation: Material Safety Data Sheet Code Number 3C016. Accessed 10 March 2009. http://www.uss.com/corp/products/msds/3c016.pdf

Van Maanen J., De Vaan M., Veldstra A. and Hendrix W. 2001. "Pesticides and Nitrate in Groundwater and Rainwater in the Province of Limburg in the Netherlands." *Environmental Monitoring and Assessment* **72**: 95–114.

Wright J., 12 November 2008. Personal Communication. Manager-Ace Roofing Company.

Yaziz M., Gunting H., Sapari N., and Ghazali A. 1989. "Variations in Rainwater Quality From Roof Catchments." *Water Research* **23**(6): 761-765.

Appendix

Table A1. Summary of Contractor Information

Residential Roofing Contractor Info	Location	Phone	Reference
Ace Roofing Company LLC	Austin, TX	(512) 836-7663	NRCA*
All-Tex Roofing Corp.	Houston, TX	(713) 683-6775	NRCA
AmeriWest	Austin, TX	(512) 287-2130	NRCA
Austin Roofing and Siding	Austin, TX	(512) 372-8110	NRCA
Barker Roofing LP	Austin, TX	(512) 243-5244	NRCA
Barr Roofing Co	Abilene, TX	(325) 672-8417	MRCA
	San Antonio,	(210) 341-3100 or	
Beldon Roofing Company	TX	(800) 688-7663	MRCA**
	San Antonio,		
Bentley Sheet Metal & Roofing Co Inc	TX	(210) 434-4184	MRCA
Billy Parker Roofing LLC	Amarillo, TX	(806) 352-1038	NRCA
	Mansfield,	(817) 477-3436 or	
Boyd Inc	TX	(888) 269-3462	MRCA
Brinkmann Roofing Co	Houston, TX	(281) 486-1660	MRCA
BRM Roofing & Construction Services			
Inc.	Houston, TX	(281) 820-8647	NRCA
Campos Roofing & Construction	El Paso, TX	(915) 755-3916	NRCA
Capco Inc.	Austin, TX	(512) 251-9516	NRCA
	Nacogdoches,		
Carney Roofing Co. Inc.	TX	(936) 569-8241	NRCA
		(214) 381-8108 or	
Castro Roofing of Texas LP	Dallas, TX	(800) 759-1879	MRCA
CBS Roofing Service	Denton, TX	(940) 387-7568	MRCA
	Longview,		
Curtis McKinley Roofing and S/M Inc.	TX	(903) 757-7402	NRCA
D.L. Phillips Construction Co. Inc.	Austin, TX	(325) 672-0204	NRCA
Daniels Roofing Inc	El Paso, TX	(915) 772-6000	MRCA
	San Antonio,		
Demarco Exteriors Plus, Inc.	TX	(210) 494-7587	MRCA
Disk Enterprises	Pearland, TX	(281) 485-7575	NRCA
Drury Roofing & Sheet Metal Inc.	Austin, TX	(512) 836-0634	NRCA
Empire Roofing	El Paso, TX	(915) 351-3550	MRCA
	Fort Worth,		
Empire Roofing Ltd	TX	(817) 572-2250	MRCA
Energy Waterproofing & Roofing			
Systems Inc.	Spring, TX	(281) 376-5171	NRCA
Escalante Enterprises	El Paso, TX	(915) 860-2672	NRCA
Frazier Roofing & Guttering Co	Arlington, TX	(817) 261-1480	MRCA
Frontier/Scholten Roof Service	El Paso, TX	(915) 845-4151	NRCA
Fry Roofing Inc	San Antonio,	(830) 980-8103	MRCA

	TX		
	Corpus		
Haeber Roofing Company	Christi, TX	(361) 851-8142	MRCA
Hamilton Roofing Company	Lubbock, TX	(806) 763-9375	MRCA
Harris Roofing Company	Mission, TX	(956) 638-0979	NRCA
	San Angelo,		
Harrison Roofing Co. Inc.	TX	(325) 653-6786	NRCA
	Longview,		
Hayes Miller Roofing Inc.	TX	(903) 758-2797	NRCA
HSR Construction Inc.	Austin, TX	(512) 458-4101	NRCA
Hynes Services Inc.	Rockport, TX	(361) 729-7180	NRCA
J. J. Flores Roofing & Construction	Laredo, TX	(956) 722-7688	NRCA
	Wichita Falls,		
Jabeau Roofing	TX	(940) 723-1183	NRCA
Jay-Co Sheet Metal and Roofing Inc.	Houston, TX	(713) 738-4525	NRCA
J-Conn Roofing & Repair Service Inc.	Austin, TX	(512) 479-0510	NRCA
	Texas City,		
John A. Walker Roofing Inc.	TX	(409) 935-5411	NRCA
John Bacon Roofing	Rockwall, TX	(972) 772-1999	MRCA
Johnson Roofing Inc	Waco, TX	(254) 662-5571	MRCA
KENTEX Roofing Systems LLC	Austin, TX	(512) 491-9000	MRCA
Long Horn Remodeling & Roofing	El Paso, TX	(915) 875-4603	NRCA
Lydick-Hooks Roofing Co of Lubbock			
Inc	Lubbock, TX	(806) 765-5577	MRCA
Lydick-Hooks Roofing Company of	Brownwood,		
Brownwood Texas Inc.	TX	(325) 646-9581	NRCA
Lydick-Hooks Roofing Company of	Wichita Falls,		
Wichita Falls Inc.	TX	(940) 322-6991	NRCA
	Wichita Falls,		
Marant Construction Inc.	TX	(940) 761-1717	NRCA
Nations Roof Central	Garland, TX	(972) 278-9200	MRCA
Norton Roofing & Construction	Amarillo, TX	(806) 372-7663	NRCA
Oliver Roofing Systems	Austin, TX	(512) 834-7500	NRCA
Parsley's S/M & Roofing Co. Inc.	Pampa, TX	(806) 669-6461	NRCA
Perry Roofing Company	Austin, TX	(512) 794-0707	NRCA
Raintree Roofing Inc.	Midland, TX	(432) 570-1822	NRCA
Rhynehart Roofing	Amarillo, TX	(806) 622-0090	NRCA
	San Angelo,		
Robles Roofing, LLC	TX	(325) 655-2416	NRCA
Roofs by Nicholas Inc.	Midland, TX	(432)520-7348	NRCA
Sechrist-Hall Company	Harlingen, TX	(956) 423-7086	MRCA
	Corpus		
Sechrist-Hall Company	Christi, TX	(361) 884-5264	MRCA
Signature Exteriors LLC	Austin, TX	(512) 481-1888	NRCA
SLR Roofing Systems Inc	Fort Worth,	(817) 731-2001	MRCA

	TX		
	Brownwood,		
Smith Roofing Co Inc	TX	(325) 646-7516	MRCA
	Brownwood,		
Smith Roofing Company Inc.	TX	(325) 646-7516	NRCA
Storm Master Inc and SMI Commercial	Fort Worth,		
Roofing	TX	(817) 589-7190	MRCA
		(512) 926-3940 or	
Texas Fifth Wall Roofing Systems Inc	Austin, TX	(800) 749-8293	MRCA
	Richardson,		
Texas Roof Management Inc	TX	(972) 272-7663	MRCA
Tower Roofing LLC	Baytown, TX	(832) 695-9474	NRCA
Vega Roofing Co.	McAllen, TX	(956) 686-4921	NRCA
Wilson Roofing Co. Inc.	Austin, TX	(512) 263-3157	NRCA

Highlighted contractors participated in survey *National Roofing Contractor Association (NRCA) **Midwest Roofing Contractor Association (MRCA)

 Table A2. Summary of Survey Questions and Answers

1. What residential roofing materials do you most commonly use (i.e., shingles, tiles, metal)?

-<u>Residential roofers</u>: asphalt fiberglass shingles.

-<u>Commercial roofers</u> did not answer this question, did not have expertise in residential roofing.

2. In your experience, what residential roofing materials are most commonly used in Texas?

-<u>Commercial and residential roofers</u>: asphalt fiberglass shingles.

3. Who manufactures these roofing materials? This information will be useful in case we want to purchase these materials for our pilot roofs.

-Commercial and residential roofers: Johns Manville, Tamko, GAF/ELK, Owens Corning,

USS, Dofasco, Capitol Roofing Company, Kemko, MonierLifetile, Roofing Supply Group, Bradco Supply.

4. Is there a regional record of what roofing materials are used? (online database or written manual)

<u>Commercial and residential roofers:</u> information is not available; contact manufacturers or roofing associations: National Roofing Contractors Association, Roofing Contractors Association of Texas, Western States Roofing Contractors Association, Tile Roofing Institute.

5. What products are used to adhere, seal, or coat roofing materials?- Commercial and residential roofers: nails and self-adhesive products.

6. If you know that the roof will be used for rainwater harvesting, what roofing materials are used?

-<u>Commercial and residential roofers:</u> anything besides asphalt based (shingles) and coal tar pitch products.

7. Is there a roofing material or adhesive that you think SHOULD NOT be used in rainwater harvesting because of its toxic nature?

-<u>Commercial and residential roofers</u>: some said that asphalt-based (shingles) and coal tar pitch products should not be used.

8. What roofing system would you recommend for rainwater harvesting because of its limited toxic materials?

-<u>Commercial and residential roofers:</u> more than 80% recommended metals; others said tiles or PVC.

*32 out of 71 contractors participated: 23 residential and 9 commercial roofing contractors

Table A3. Synthetic Organic Compounds Tested

1,2,4-Trichlorobenzene 1,2-Dichlorobenzene 1,3-Dichlorobenzene 1,4-Dichlorobenzene 2,4,5-Trichlorophenol 2,4,6-Trichlorophenol 2,4-Dichlorophenol 2,4-Dimethylphenol 2,4-Dinitrophenol 2,4-Dinitrotoluene 2,6-Dichlorophenol 2,6-Dinitrotoluene 2-Chloronaphthalene 2-Chlorophenol 2-Methylnaphthalene 2-Methylphenol 2-Nitroaniline 2-Nitrophenol 3,3'-Dichlorobenzidine 3-Nitroaniline 4,6-Dinitro-2-methylphenol 4-Bromophenyl phenyl ether 4-Chloro-3-methylphenol 4-Chloroaniline 4-Chlorophenyl phenyl ether 4-Methylphenol 4-Nitroaniline 4-Nitrophenol Acenaphthene Acenaphthylene Aniline Anthracene Benzo[a]anthracene Benzo[a]pyrene Benzo[b]fluoranthene Benzo[g,h,i]perylene Benzo[k]fluoranthene Benzyl alcohol Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-ethylhexyl)phthalate Butyl benzyl phthalate Carbazole

Chrysene Di-n-butyl phthalate Di-n-octyl phthalate Dibenz[a,h]anthracene Dibenzofuran Diethyl phthalate Dimethyl phthalate Fluoranthene Fluorene Hexachlorobenzene Hexachlorobutadiene Hexachlorocyclopentadiene Hexachloroethane Indeno[1,2,3-cd]pyrene Isophorone N-Nitrosodi-n-propylamine N-Nitrosodiethylamine N-Nitrosodiphenylamine Naphthalene Nitrobenzene Pentachlorophenol Phenanthrene Phenol Pyrene Surr: 2,4,6-Tribromophenol Surr: 2-Fluorobiphenyl Surr: 2-Fluorophenol Surr: 4-Terphenyl-d14 Surr: Nitrobenzene-d5 Surr: Phenol-d6 21.5 0 20 -TIC: 2,2-Dimethyl-6-cyclohexanepropanol, TIC: Ethyl cyclopropanecarboxylate, 1,1,1,2-Tetrachloroethane 1,1,1-Trichloroethane 1,1,2,2-Tetrachloroethane 1,1,2-Trichloroethane 1,1-Dichloroethane 1,1-Dichloroethene 1,1-Dichloropropene 1,2,3-Trichlorobenzene 1,2,3-Trichloropropane 1,2,4-Trichlorobenzene 1,2,4-Trimethylbenzene 1,2-Dibromo-3-chloropropane 1,2-Dibromoethane 1.2-Dichlorobenzene

1,2-Dichloroethane 1,2-Dichloropropane 1,3,5-Trimethylbenzene 1,3-Dichlorobenzene 1,3-Dichloropropane 1,4-Dichlorobenzene 2,2-Dichloropropane 2-Chlorotoluene 4-Chlorotoluene Benzene Bromobenzene Bromochloromethane Bromodichloromethane Bromoform Bromomethane Carbon tetrachloride Chlorobenzene Chloroethane Chloroform Chloromethane cis-1,2-Dichloroethene cis-1,3-Dichloropropene Dibromochloromethane Dibromomethane Dichlorodifluoromethane Ethylbenzene Hexachlorobutadiene Isopropylbenzene m,p-Xylene Methylene chloride n-Butylbenzene n-Propylbenzene Naphthalene o-Xylene p-Isopropyltoluene sec-Butylbenzene Styrene tert-Butylbenzene Tetrachloroethene Toluene trans-1,2-Dichloroethene trans-1,3-Dichloropropene Trichloroethene Trichlorofluoromethane Vinyl chloride Surr: 1,2-Dichloroethane-d4 Surr: 4-Bromofluorobenzene Surr: Dibromofluoromethane

MATERIAL SAFETY DATA SHEET

MSDS Number: T029000

Revision Date: May 2007

1. CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

TRADE NAME: Shingles LABEL: TAMKO USE & DESCRIPTION: Roofing Shingles CHEMICAL FAMILY: Mixture

MANUFACTURED FOR:
TAMKO Building Products, Inc.EMERGENCY TELEPHONE NUMBERS;
General Information: 1-417-624-6644 (8 a.m. - 5 p.m. CST)
Chemtrec: 1-800-424-9300Joplin, MO 64802-1404Chemtrec: 1-800-424-9300

2. COMPOSITION/INFORMATION ON INGREDIENTS

			Exposure Limits*				
Components	Cas No.	% by Wt.	OSHA	OSHA ACGIH			
			TWA	STEL	TWA	STEL	Unit
Petroleum asphalt	8052-42-4	<30	5 fume	NE	0.5 <mark>f</mark> ume	NE	mg/m ³
Limestone**	1317-65-3	<65	10 resp dust	NE	10 resp dust	NE	mg/m ³
Mineral Granules	NE	<40	NE	NE	NE	NE	
MAT Fiber Glass Urea Formaldehyde Binder	65997-17-3 9011-05-6	<8 <2.4	NE	NE	1 fiber	NE	сс
Formaldehyde Polyester Felt	50-00-0 NE NE	<0.1	0.75 ppm NE NE	2 ppm NE NE		0.3ppm NE NE	ppm
BACKING Sand** Talc**	14808-60-7 4807-96-6	<10		NE NE		NE NE	mg/m ³ mg/m ³
** contains: crystalline silica >5% quartz crystobalite	14808-60-7 14464-46-1	>0.1%	.05 resp dust .05 resp dust	NE NE	.05 resp dust .05 resp dust	NE NE	mg/m ³ mg/m ³

NE = Not established

* Note: Due to the form of the product, hazardous exposures are not expected to occur. Exposure limits are provided for information purposes only.

DOFASCO



Material Safety Data Sheet

Motor	iol	ZINC COATE		- PRODUCT IDE	NTIFICATION	WHMIS Class
Mater Synor					na Diva Zinaramatal	
-	facturer				me Plus, Zincrometal io, Canada L8N 3J5	D2A, D2B
eiep	hone No.	(905) 548-720	0	Material Use	Manufacture of steel art	
				- HAZARDOUS		
	dous Ingredi	ents	Weight %	CAS No.	LD50	Exposure Limit (mg/m ³)
Steel:	1		05	7400.00.0	20 - (line (met a mell)	F (F ,, c)
	Iron (Fe)	N (m)	~ 95	7439-89-6	30 g/kg (rat-oral) 9 g/kg (rat-oral)	5 (Fume)
	Manganese (≤ 1.65 ≤ 1.1	7439-96-5	9 g/kg (rat-orai) U	0.2 0.5
	Chromium (C Nickel (Ni)	a)	≤ 1.1 ≤ 0.12	7440-47-3 7440-02-0	U	1.5
	(Hazardous I	ngredients – lists o	components wh	nich meet the reportin	ng requirements of the Hazar	rdous Products Act.)
coating	•					
	Galvanized					
	Zinc (Zn)	o	99	7440-66-6	U	5 (Fume)
	(Z-coating.	Coating weights	range from 1	5 to 500 g/m ⁻ or u	p to 20% total steel weigh	t)
	Galvanneal					
	Zinc (Zn)		88	7440-66-6	U	5 (Fume)
	Iron (Fe)		11	7439-89-6	Ŭ	5 (Fume)
		-coating. Coatir			g/m ² or up to 10% total s	
3.	Galvalume (alvalume Plus				
	Aluminum (A		55	7429-90-5	U	10
	Zinc (Zn)	·/	43	7440-66-6	Ŭ	5 (Fume)
		Coating weight			up to 15% total steel)	e (Fune)
lurface	e Treatments;					
Sunace						
	(Constitutes I	ess than 0.5% of t	otal steel weig	ht)		
					n residual of 11 to 40mg/n	n ² per side.
	2. Slushin	g Oil - (Ferroco	te 61 MAL H	CL-1G, Ferrocote	61-AUS, PL-7105-A)	
						onates and anti-oxidants.
					affin petroleum distillate.	
				5.4 g/m ² per side.		
	4. Pre Tei Gal	nper - (Qwerl 26 valume Plus – (C	oo-∟∨) White Dakite PC461	petroleum mineral 0) Acrylic coating (oil. of polystyrene-acrylate co	polymers.
Note:					tact Dofasco's Technical Se	
.0.0.						one depleting substances.

Legend: U = Unknown NA = Not Applicable

MonierLifetile MSDS - Material Safety Data Sheet



SECTION I - PRODUCT AND COMPANY IDEN	TIFICATION
Identity (As used on Label and List)	Trade Names
MONIERLIFETILE CONCRETE ROOF TILES	Cedarlite, Duralite, Standard Weight Tile and Trim
Manufacturer's Name	Emergency Telephone Number
MonierLifetile, LLC	(949) 981-3319
Address (Number, Street, City, State, and ZIP Code) MonierLifetile, LLC	Telephone Number for Information (800) 224-2024 ext.
7575 Irvine Center Drive, Suite 100	Date Prepared
Irvine, CA 92618 - 2930	April 20, 1999 Revised : February 17, 2004, Revision G

Hazardous Components/CAS Number	OSHA PEL (mg/m3)	ACGIH TLV (mg/m3)	% by weight
Portland Cement 65997-15-1	15 Total 5 Respirable	10 Total 3 Respirable	20-30%
Sand and Aggregate (variable crystalline Silica content) 14808-60-7	15 Total 5 Respirable	10 Total 3 Respirable	50-60%
Limestone 1317-65-3	15 Total 5 Respirable	10 Total 3 Respirable	0-5%
Fly Ash 68131-74-8	15 Total 5 Respirable	10 Total 3 Respirable	0-8%
Mold Release Agent (diesel/petroleum oil, vegetable oil)	None Established For Vapor	None Established For Vapor	Less than 1%
Acrylic Polymer	None Established	None Established	0-8%
Metal Oxide Pigments (various mixtures to produce color section): Cobalt Metal Pigments (blue) 1307-96-6 Iron Oxide Pigments (black, red and yellow) 1309-37-1 Titanium Dioxide Pigment (white) 13463-67-7 Chromium (111) Oxide Pigments (green) 1308-38-9 Aquis Dispersions	 0.1 10 (fume) 15 1 	 0.05 5 (dust and fume) 10 0.5 	0 - 3.0% < 1.0% 0 - 3.0% 0 - 3.0% 0 - 3.0% 0 - 3.0%

Boiling Point	Specific Gravity (H2O = 1)
Not applicable	Denser than water
Vapor Pressure (mm Hg.)	Melting Point
Not applicable	Not applicable
Vapor Density (AIR = 1)	Evaporation Rate (Butyl Acetate = 1)
Not applicable	Not applicable
Solubility in Water Negligible	· · · · · · · · · · · · · · · · · · ·